## STUDENT A

- 1. Read the text below and try to complete the missing words.
- 2. Listen to student B giving you the definitions of the missing words and fill the gaps.
- 3. Explain the words in bold to student B.

## **OXYGEN**

Adapted from <a href="http://en.wikipedia.org/wiki/Oxygen">http://en.wikipedia.org/wiki/Oxygen</a>

Oxygen is the element v	vith <b>1.</b>	and repres	sented by the symbol <b>O</b> . At
standard temperature a	and pressure, two ato	oms of the element	bind to form dioxygen,
a colorless, 2	, tasteless <b>3.</b>	gas with	the formula O <sub>2</sub> . Oxygen is
a member of the <b>4.</b>	group on	the <b>5.</b>	, and is a highly
<b>6.</b> nonme	etallic element that rea	dily forms 7	(notably oxides)
with almost all other el	lements. By mass, oxy	gen is the third mos	at abundant element in the
universe after hydrogen	n and <b>8.</b> a	and the most abundar	nt element by mass in the
9	, making up almos	st half of the crust's	mass. Free oxygen is too
chemically reactive to a	appear on Earth withou	ut the <b>10.</b>	action of
living organisms, which	use the energy of sur	nlight to produce eler	nental oxygen from water.
Elemental O2 only bega	n to accumulate in the	e atmosphere after th	e evolutionary appearance
of these organisms, rou	ghly 2.5 billion years	ago. Diatomic oxyge	n gas constitutes 20.8% of
the <b>11.</b> of air	r.		

Oxygen comprises most of the mass of living organisms (for example, about 12.two-thirds of the human body's mass). All major classes of structural molecules in living organisms, such as 13. proteins, 14. carbohydrates, and fats contain oxygen, as do the major 15. inorganic compounds that comprise animal shells, teeth, and bone. Elemental oxygen is produced by cyanobacteria, algae and plants, and is used in 16.cellular respiration for all complex life. Oxygen is 17.toxic to obligately anaerobic organisms, which were the dominant form of early life on Earth until  $O_2$  began to accumulate in the atmosphere. Another form (allotrope) of oxygen, 18. ozone  $(O_3)$ , helps protect the biosphere from 19. ultraviolet radiation with the high-altitude 20. ozone layer, but is a pollutant near the surface where it is a 21. by-product of smog. At even higher low earth orbit 22. altitudes atomic oxygen is a significant presence and a cause of erosion for spacecraft.

## STUDENT B

- 1. Read the text below and try to complete the missing words.
- 2. Explain the words in bold to student A.
- 3. Listen to student A giving you the definitions of the missing words and fill the gaps.

## **OXYGEN**

Adapted from <a href="http://en.wikipedia.org/wiki/Oxygen">http://en.wikipedia.org/wiki/Oxygen</a>

Oxygen is the element with **1. atomic number 8** and represented by the symbol **O**. At standard temperature and pressure, two atoms of the element bind to form dioxygen, a colorless, **2. odorless**, tasteless **3. diatomic** gas with the formula O<sub>2</sub>. Oxygen is a member of the **4. chalcogen** group on the **5. periodic table**, and is a highly **6. reactive** nonmetallic element that readily forms **7. compounds** (notably oxides) with almost all other elements. By mass, oxygen is the third most abundant element in the universe after hydrogen and **8. helium** and the most abundant element by mass in the **9. Earth's crust**, making up almost half of the crust's mass. Free oxygen is too chemically reactive to appear on Earth without the **10. photosynthetic** action of living organisms, which use the energy of sunlight to produce elemental oxygen from water. Elemental O<sub>2</sub> only began to accumulate in the atmosphere after the evolutionary appearance of these organisms, roughly 2.5 billion years ago. Diatomic oxygen gas constitutes 20.8% of the **11. volume** of air.

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