## **FIREWORKS**

http://www.youtube.com/watch?v=0Mbf1Pfl\_2U

## 1. Watch the film and choose the right answer:

a.	Potassium nitrate is used:		
1)	to produce red colour		
2)	as a fuel		
3)	to make black powder		
b.	is used in a lot	of c	olour compositions
	Sulfur	<i>31</i> C	olour compositions.
	Potassium perchlorate		
	-		
3)	Magnesium powder		
c.	To produce heat we need:		
1)	only a fuel		
2)	a fuel and an oxidizer		
3)	an aerial shell and sulfur		
4	The eariel shall deepn't include:		
	The aerial shell doesn't include:		
	a black powder propellant		
	a time fuse		
3)	strontium nitrate		
2.	<b>Complete the sentence:</b>		
We produce colours using the effect that different chemicals heated at get rid			
	this energy meeting very specific		
	6,		
3.	Match the substances with	th	e colour that they produce:
a.	copper oxide	1.	yellow
b.	strontium chloride	2.	blue
c.	sodium silicate	3.	green
d.	calcium nitrate	4.	red
e.	barium acetate	5.	yellow

## 4. True or false?

- a. To produce colour you start with a burning mixture.
- b. A burning mixture contains a fuel and a colour component.
- c. Blue is the hardest colour to produce.
- d. A red flame composition from strontium nitrate has a very bright flame because it produces heat.
- e. A white sparks effect is produced by a base mixture and potassium nitrate.